## Science 8 - Density Calculations Worksheet



1) A student measures the mass of an $8 \mathrm{~cm}^{3}$ block of brown sugar to be 12.9 g . What is the density of the brown sugar? $m=12.99$

$$
\begin{aligned}
& m=12.99 \\
& v=8 \mathrm{~cm}^{3}
\end{aligned}
$$



$$
D=1.6125 \mathrm{~g} / \mathrm{cm}^{3}
$$

2) A chef fills a 50 mL container with 43.5 g of cooking oil. What is the density of the oil?
$V=50 \mathrm{~mL}$
$D=\frac{m}{V}$
$D=\frac{43.5 \mathrm{~g}}{50 \mathrm{~mL}}$
$D=.87 \mathrm{~g} / \mathrm{mL}$
3) Calculate the mass of a liquid with a density of $2.5 \mathrm{~g} / \mathrm{mL}$ and a volume of 15 mL .

$$
\begin{aligned}
& D=2.5 \mathrm{~g} / \mathrm{mL} \\
& V=15 \mathrm{~mL}
\end{aligned} \quad 2.5 \mathrm{~g} / \mathrm{mL}=\frac{m}{15 \mathrm{~mL}} \quad m=(2.5)(15) \quad m=37.5 \mathrm{~g}
$$

4) Calculate the volume of a liquid with a density of $5.45 \mathrm{~g} / \mathrm{mL}$ and a mass of 65 g .
$D=5.45 \mathrm{~g} / \mathrm{mL}$
$m=65 \mathrm{~g}$
$5.45=\frac{65}{V} \quad V=\frac{65}{5.45}$
$V=11.93 \mathrm{~mL}$
5) A machine shop worker records the mass of an aluminum cube as 176 g . If one side of the cube measures 4 cm , what is the density of the aluminum?
$m=176 \mathrm{~g}$
$\begin{aligned} V & =L \times \omega \times h \\ & =4 \times 4 \times 4\end{aligned}>V=64 \mathrm{~cm}^{3}$
$D=\frac{1769}{64 \mathrm{~cm}^{3}}$
$D=2.75 \mathrm{~g} / \mathrm{cm}^{3}$
6) A teacher performing a demonstration finds that a piece of cork displaces 23.5 mL of water. The piece of cork has a mass of 5.7 g . What is the density of the cork?
$m=5.79$
$D=\frac{5.7}{23.5}$
$D=.24 \mathrm{~g} / \mathrm{mL}$
7) A carver begins work on the following block of granite that weighs 2700 g . What is the density of the granite?

8) A piece of PVC plumbing pipe displaces 60 mL when placed into a container of water. If the pipe has a mass of 78 g , what is the density of PVC?

$$
\begin{aligned}
& V=60 \mathrm{~mL} \\
& m=789 \quad D=\frac{78}{60} \quad D=1.3 \mathrm{~g} / \mathrm{mL}
\end{aligned}
$$

9) A solid magnesium flare has a mass of 1300 g and a volume of $743 \mathrm{~cm}^{3}$. What is the density of the magnesium?

$$
\begin{aligned}
& M=1300 \mathrm{~g} \\
& V=743 \mathrm{~cm}^{3}
\end{aligned} \quad D=\frac{1300}{743}
$$


10) A graduated cylinder has a mass of 50 g when empty. When 30 mL of water is added, the graduated cylinder has a mass of 120 g . If a rock is added to the graduated cylinder, the water level rises to 75 mL and the total mass is now 250 g . What is the density of the rock?

$$
\begin{aligned}
V & =75 \mathrm{~mL}-30 \mathrm{~mL} \\
& =45 \mathrm{~mL}
\end{aligned}
$$

$$
120 \mathrm{~g}-50 \mathrm{~s}
$$

(2) 250-50

11) A student performs an experiment with three unknown fluids and obtains the following measurements:

Fluid A: $m=2060 \mathrm{~g}, \mathrm{~V}=2000 \mathrm{~mL}$
Fluid B: $m=672 \mathrm{~g}, \mathrm{~V}=850 \mathrm{~mL}$
Fluid C: $m=990 \mathrm{~g}, \mathrm{~V}=1100 \mathrm{~mL}$
Draw how the fluids would be layered if they were combined in a beaker.

$$
\text { D. } \begin{array}{rlrl}
A & =\frac{2060}{2000} & D B= & =\frac{672}{850} \\
& D C & =\frac{990}{1100} \\
& =1.03 & =.79 & =.9
\end{array}
$$


12) Use your density skills to find the identity of the following mystery objects.

| Table of Densities |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Solids | Density $\mathbf{g} / \mathbf{c m}^{\mathbf{3}}$ | Solids | ${\text { Density } \mathbf{g} / \mathbf{c m}^{\mathbf{3}}}^{\|c\|}$ Copper |  |
| Marble | 2.56 | Gold | 8.92 |  |
| Quartz | 2.64 | Platinum | 19.32 |  |
| Diamond | 3.52 | 21.4 |  |  |



What is the coin made of? Copper

You find a ring with a mass of 107 g . You fill a graduated cylinder up with 10 mL of water and put the ring into the cylinder. The water rises up to the 15 mL mark.

$$
\begin{aligned}
m=107 \mathrm{~g} \quad V & =15 \mathrm{~mL}-10 \mathrm{~mL} \\
& =5 \mathrm{~mL} \\
D=107 / 5 & =21.4 \mathrm{P}
\end{aligned}
$$

What is the ring made of? $\qquad$


You think you have found a diamond. Its mass is 5.28 g and its volume is $2 \mathrm{~cm}^{3}$.

$$
D=\frac{5.28}{2}=2.64
$$

What did you $\qquad$ Quartz

There is a block on your desk that acts as a paperweight. Its measurements are 3 cm by 4 cm by 6 cm . The block has a mass of 184.32 g .

$$
\begin{array}{rlrl}
V & =L \times w \times h \quad m=184.32 \mathrm{~g} & D & =\frac{184.32}{72} \\
& =3 \times 4 \times 6 & & D=2.56 \\
& =72 \mathrm{~cm}^{3} & &
\end{array}
$$

What is the block made of? Marble

