Test 3 Extra Synthesis Practice Problems
Page 1: Synthesis Design Practice.
Page 2 +3 : Predict the Product Practice (including some that involve stereochemistry).
Page 4: Cis/trans Stereospecific reactions: which recipe to use; which E or Z alkene to use.
Page 5: Recognizing cationic/anionic/radical reactions, and reasonable intermediates/first steps
Page 6: Elements of unsaturation/hydrogenation problems; ozonolysis puzzle problems.
A. Provide reagents for the following transformations.
1.


3.


4.


1. $\mathrm{NE}_{3}$ (bulky base) 2. $\mathrm{BH}_{3} \cdot \mathrm{THF}$

2. $\mathrm{NaOH} \mathrm{H}_{2} \mathrm{O}_{2}$

3. 



1. $\mathrm{Br}_{2}, \mathrm{hv}$
2. NaOH (small base)

3. $\mathrm{O}_{3}$
4. $\mathrm{Me}_{2} \mathrm{~S}$

5. $\mathrm{Br}_{2}, h u$
6. NaOH
7. $\mathrm{CH}_{3} \mathrm{CO}_{3} \mathrm{H}, \mathrm{H}_{2} \mathrm{O}$
8. 

B. Draw the major product for each of the following reactions or reaction sequences. You needn't bother to show side products or minor products. For chiral molecules that are racemic, you needn't draw both enantiomers. BE CAREFUL TO SHOW THE CORRECT ORIENTATION, AND THE CORRECT STEREOCHEMISTRY IN CASES WHERE STEREOCHEM IS FACTOR. (3 points each).
9.

.

10.

11.

12.


13.

14.



1. $\mathrm{Hg}(\mathrm{OAc})_{2}, \mathrm{H}_{2} \mathrm{O}$
2. 
3. $\mathrm{NaBH}_{4}$

 (via

4. 
5. $\mathrm{H}_{2} \mathrm{SO}_{4}$
6. 



Note: explicit stereochemistry must be drawn. The enantiomer would have been equally acceptable.
$\leq \quad 21$.

20.

 Cuia

22.


23.


Note: explicit stereochemistry must be drawn. The enantiomer would have been equally acceptable. This principle will apply for any of the reactions producing two chiral centers. Problems 23-32
25.

27.
26.


28.
29.

30.
31.




32.

33.


34.

35.

C. Draw the alkene that would product the products shown. Make sure to make your drawing clear whether the starting alkene was E or Z .

37.

38.

39.



41.

oops
D. What reagent(s) would you use to conduct the following transformations?
42.

43.

E. Recognizing whether reaction mechanisms should be cationic, anionic, or radical; whether intermediates should be cationic, anionic, or radical; and recognizing what could be reasonably involved in the initial reaction step.
44. The transformation shown is common in many biological systems. Which of the following statements is definitely, absolutely false?
a. The first step in the mechanism probably involves protonation of the carbonyl oxygen.

b. The overall reaction involves an addition T

## c. The mechanism is probably radical in nature Definitely false

reaction
c. The mechanism is probably radical in nature
45. For the transformation shown, which of the
following statements is definitely, absolutely
reaction
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45. For the transformation shown, which of the
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c. The mechanism is probably radical in nature
45. For the transformation shown, which of the
following statements is definitely, absolutely false?
a.) The first step in the mechanism probably
involves protonation of a carbonyl oxygen.



b. The overall reaction involves a substitution $T$ reaction
c. The mechanism is probably anionic in nature $T$
d. The first step in the mechanism involves
ethoxide anion grabbing a hydrogen.
46. Shown is a reaction, and some possible intermediates along the mechanistic pathway. Given the reaction conditions shown, which of the following statements is true?
a. Structures $\mathbf{A}$ and $\mathbf{B}$ might be plausible intermediates; structure $\mathbf{C}$ definitely isn't
b. Structures $\mathbf{A}$ and $\mathbf{C}$ might be plausible intermediates; structure $\mathbf{B}$ definitely isn't
c. Structures $\mathbf{B}$ and $\mathbf{C}$ might be plausible intermediates; structure A definitely isn't
d. Structure A might be a plausible

intermediates; structures $\mathbf{B}$ and $\mathbf{C}$ definitely aren't
47. Shown is a reaction, and some possible intermediates along the mechanistic pathway. Given the reaction conditions shown, which of the following statements is true?
a. Structures $\mathbf{A}$ and $\mathbf{B}$ might be plausible intermediates; structure $\mathbf{C}$ definitely isn't
b. Structures $\mathbf{A}$ and $\mathbf{C}$ might be plausible intermediates; structure $\mathbf{B}$ definitely isn't
c. Structures B and C might be plausible intermediates; structure A definitely isn't

d. Structure A might be a plausible intermediates; structures $\mathbf{B}$ and $\mathbf{C}$ definitely aren't
F. Elements of Unsaturation/Hydrogenation Problems. For each problem there will be multiple satisfactory solutions.
48. Provide a possible structure for a compound with formula $\mathrm{C}_{5} \mathrm{H}_{8}$, given that it reacts with excess $\mathrm{H}_{2} / \mathrm{Pt}$ to give $\mathrm{C}_{5} \mathrm{H}_{10}$.

Answer must show one alkene and one ring. (Other structures also meet that requirement). $\mathrm{H} 2 / \mathrm{Pt}$ test proved 1 alkene.
$\mathrm{EU}=2$ originally.

etc
49. Provide a possible structure for a Answer must show two alkene and one ring. compound with formula $\mathrm{C}_{6} \mathrm{H}_{8}$, given (Other structures also meet that requirement). that it reacts with excess $\mathrm{H}_{2} / \mathrm{Pt}$ to $\mathrm{H} 2 / \mathrm{Pt}$ test proved 2 alkenes. $\mathrm{EU}=3$ originally.
 give $\mathrm{C}_{6} \mathrm{H}_{12}$.

So the other EU must be ring.

50. Provide a possible structure for a Answer must show two alkenes and two rings. compound with formula $\mathrm{C}_{8} \mathrm{H}_{10}$, given that it reacts with excess $\mathrm{H}_{2} / \mathrm{Pt}$ (Other structures also meet that requirement). $\mathrm{H} 2 / \mathrm{Pt}$ test proved 2 alkene. $\mathrm{EU}=4$ originally.
 to give $\mathrm{C}_{8} \mathrm{H}_{14}$.

So the other two EU must be two rings.
51. Provide possible structure for a compound with formula $\mathrm{C}_{6} \mathrm{H}_{\text {en }}$ given

Answer show two alkene and one ring. that it reacts with excess $H_{2} / \mathrm{P}+$ to $\mathrm{H} 2 / \mathrm{Pt}$ test proved 2 alkenes.
$-\mathrm{C}_{6} \mathrm{H}_{12}$.
$\mathrm{EU}=3$ originally.

Oops, same as \# 49
G. Ozonolysis: Draw starting chemicals that will undergo ozonolysis to produce the products shown. In some cases there may be more than one satisfactory answer.
52.


53.





